Nuclear Check



2) What is the mass number of an alpha particle?

1) 1 2) 2 3) 0 4) 4

- 3) When an atom of the unstable isotope Na-24 decays, it becomes an atom of Mg-24 because the Na-24 atom spontaneously releases
 - 1) an alpha particle 3) a neutron
 - 2) a beta particle 4) a positron
- 4) Which particle is emitted from a hydrogen-3 nucleus when it undergoes radioactive decay?
 - β⁻ 3) B+ 4) Y 1) α
- 5) What is the decay mode of 37 K?
 - 3) γ β⁻ 2) β⁺ 4) a
- 6) Which equation represents positron decay?
 - 1) ${}^{87}_{37}\text{Rb} \rightarrow {}^{0}_{-1}\text{e} + {}^{87}_{38}\text{Sr}$ 2) ${}^{277}_{92}U \rightarrow {}^{223}_{90}Th + {}^{4}_{2}He$ 3) $^{27}_{13}\text{Al} + ^{4}_{2}\text{He} \rightarrow ^{30}_{15}\text{P} + ^{1}_{0}\text{n}$
 - 4) ${}^{11}_{6}C \rightarrow {}^{0}_{\pm 1}e + {}^{11}_{5}B$
- 7) Which nuclear decay emission consists of energy, only?
 - 1) alpha particle 3) gamma radiation
 - 4) positron 2) beta particle
- 8) Which radioisotope has the fastest rate of decay?
 - 1) ${}^{14}C$ 2) ${}^{37}Ca$ 3) ${}^{53}Fe$ 4) ${}^{42}K$
- What is the mass of an original 5.60-gram 9) sample of iron-53 that remains unchanged after 25.53 minutes?

1)	0.35 g	3)	1.40 g
2)	0.70 g	4)	2.80 g

10)The diagram below represents radiation passing through an electric field.



Which type of emanation is represented by the arrow labeled 2?

- 1) alpha particle 3) positron
- 2) beta particle 4) gamma radiation
- 11) What fraction of a Sr-90 sample remains unchanged after 87.3 years?
 - 1) <u>1</u> 2 2) 3 3) 4) 8
- 12) What mass of a 60.0-gram sample of ¹⁶N will remain unchanged after 28.8 seconds?
 - 1) 3.75 g 3) 15.0 g 2) 7.50 g 4) 30.0 g